# Mutational Analysis of Carbamyl Phosphate Synthetase. Substitution of Glu841 Leads to Loss of Functional Coupling between the Two Catalytic Domains of the Synthetase Subunit<sup>†</sup>

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ABSTRACT: The synthetase subunit of Escherichia coli carbamyl phosphate synthetase has two catalytic nucleotide-binding domains, one involved in the activation of  $HCO_3^-$  and the second in phosphorylation of carbamate. Here we show that a Glu841  $\rightarrow$  Lys841 substitution in a putative ATP-binding domain located in the carboxyl half of the synthetase abolishes overall synthesis of carbamyl phosphate with either glutamine or  $NH_3$  as the nitrogen source. Measurements of partial activities indicate that while  $HCO_3^-$ -dependent ATP hydrolysis at saturating concentrations of substrate proceeds at higher than normal rates, ATP synthesis from ADP and carbamyl phosphate is nearly completely suppressed by the mutation. These results indicate Glu841 to be an essential residue for the phosphorylation of carbamate in the terminal step of the catalytic mechanism. The Lys841 substitution also affects the kinetic properties of the  $HCO_3^-$  activation site. Both  $k_{cat}$  and  $K_m$  for ATP increase 10-fold, while  $K_m$  for  $HCO_3^-$  is increased 100-fold. Significantly,  $NH_3$  decreases rather than stimulates  $P_i$  release from ATP in the  $HCO_3^-$ -dependent ATPase reaction. The increase in  $k_{cat}$  of the  $HCO_3^-$ -dependent ATPase reaction, and an impaired ability of the Lys841 enzyme to catalyze the reaction of  $NH_3$  with carboxy phosphate, strongly argues for interactions between the two catalytic ATP sites that couple the formation of enzyme-bound carbamate with its phosphorylation.

In Escherichia coli, carbamyl phosphate, an essential intermediate in the arginine and pyrimidine biosynthetic pathways, is synthesized by glutamine-dependent carbamyl phosphate synthetase (Pierard et al., 1965). Carbamyl phosphate is formed from glutamine (or NH<sub>3</sub>) and HCO<sub>3</sub><sup>-</sup> in a reaction requiring two molecules of ATP, only one of which donates its phosphoryl group to carbamyl phosphate (Anderson & Meister, 1965). Carbamyl phosphate synthetase is composed of two catalytic subunits (Matthews & Anderson, 1972; Trotta et al., 1971). The smaller glutaminase subunit hydrolyzes glutamine and transfers enzyme-bound NH<sub>3</sub> (eq 1 in Scheme I) to a site located on the synthetase subunit (Trotta et al., 1971, 1974b).

Scheme I

E + glutamine + 
$$H_2O \rightleftharpoons glutamate + E \cdot NH_3$$
 (1)

$$E + ATP + HCO_3^- \rightleftharpoons E \cdot CO_3PO_3^{3-} + ADP + H^+$$
 (2)

$$E \cdot CO_3 PO_3^{3-} + NH_3 \rightleftharpoons E \cdot CO_2^{-}NH_2 + P_i$$
 (3)

$$E \cdot CO_2 - NH_2 + ATP \rightleftharpoons E + carbamyl - P + ADP$$
 (4)

The mechanism of carbamyl phosphate synthesis is known to proceed through carboxy phosphate (Anderson & Meister, 1965; Raushel & Villafranca, 1980; Wimmer et al., 1979). This intermediate is formed by an initial phosphorylation of bicarbonate resulting in an enzyme-bound carboxy phosphate (eq 2). Although the enzyme-bound carboxy phosphate is not easily detected because of its lability, the rate of its breakdown can be assayed by measuring HCO<sub>3</sub>-dependent ATPase (Anderson & Meister, 1966). In the second step, the bound carboxy phosphate undergoes a nucleophilic attack by NH<sub>3</sub>, yielding enzyme-bound carbamate (eq 3). Carbamate is phosphorylated by the second molecule of ATP to form car-

bamyl phosphate in the terminal step (eq 4). The latter reaction, because its equilibrium favors ATP formation, can be assayed by measuring the synthesis of ATP from ADP and carbamyl phosphate (Anderson & Meister, 1966).

The above mechanism suggests the existence of at least three functionally distinct sites in the enzyme, each defined by its ability to bind and act on specific substrates and intermediates of the catalytic cycle. On the basis of primary sequence homology with other amidotransferases and in vitro mutagenesis studies, the region of the glutaminase with the active-site cysteine involved in glutamine hydrolysis has been identified (Nyunoya & Lusty, 1984; Rubino et al., 1986). The sequences of both glutamine- and ammonia-dependent carbamyl phosphate synthetases have also disclosed the existence of at least two nucleotide-binding sites in the synthetase component (Lusty et al., 1983). Amino acid substitutions in two of the proposed nucleotide-binding sites of E. coli carbamyl phosphate synthetase have confirmed that the initial activation of bicarbonate and the terminal phosphorylation of carbamate are functions of two separate domains, located in the aminoand carboxyl-terminal halves of the synthetase (Post et al., 1990). Even though some of the partial reactions have been successfully dissected by mutational (Post et al., 1990) as well chemical means, biochemical and, more recently, genetic data also point to the existence of reciprocal modulatory influences of reaction centers on the synthetase and the glutaminase components (Anderson & Carlson, 1975; Foley et al., 1971; Pinkus & Meister, 1972; Rubino et al., 1986, 1987). Such evidence implies either conformationally transmitted signals between active sites or, alternatively, that some of the active sites may physically overlap and share common residues.

To gain additional insights into the organization of the functional domains in carbamyl phosphate synthetase, we have undertaken a mutational analysis of the *E. coli* enzyme. The *carA* and *carB* genes of this bacterium code for the glutaminase and synthetase subunits, respectively (Mergeay et al.,

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strain	genotype	source or ref
K12	F-	CSH collection <sup>a</sup>
P4X	Hfr, metB1, relA1, spoT1	B. Bachmann <sup>b</sup>
C600	$F_{r_k}$ , $r_m$ , $\Delta car B$ -8	M. Crabeel <sup>c</sup>
RC50	F-, $carA$ -50, $thi$ -1, $malA$ 1, $xyl$ -7, $rpsL$ 135, $\lambda^R$ , $\lambda^-$ , $tsx$ -273	Mergeay et al. (1974)
L623	as RC50, but car-A50 zab-27::Tn10	Guillou et al. (1989)
RR1	F-, pro-, leuB, thi-1, lacY1, hsdR, endA, rpsL20, ara-14, galK2, xyl-5, mtl-1, supE44, $\lambda$	Bolivar & Backman (1979)
L634	as RR1, but car-50 zab-27::Tn10	$P1_{vir}(L623) \times RR1$
L649	as RR1, but carA-50, Leu+	$P1_{vir}(RC50) \times L634$
L814	as L649, but recA56, srlC-300::Tn10	JC10240 <sup>d</sup> × L649
L895	C600, but $carAB^+$ $zab-21::Tn10$	this laboratory
L683	as RR1, but carB-2117 zab-21::Tn10	this study
L705	as RR1, but carB-2117 zab-21::Tn10	this study
L829	$C600(pLLK12, carAB^+)$	this study
L871	L814(pLL43, carB-2117)	this study
L880	C600(pLL117, carAB-2117)	this study

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1974). In the present paper, we describe a mutation in carB resulting in a single amino acid substitution in the synthetase subunit that suppresses overall carbamyl phosphate synthesis by selectively blocking the terminal step of the catalytic mechanism (eq 4). Kinetic analysis of the mutant enzyme has uncovered a novel form of functional coupling between the sites involved in carboxy phosphate formation and in the conversion of carbamate to carbamyl phosphate. The identification of the residue responsible for the altered properties of the mutant synthetase subunit permits further refinements in the model of the carbamate phosphorylating domain in the carboxyl half of the protein.

## MATERIALS AND METHODS

Chemicals and Enzymes. Restriction endonucleases and other DNA-modifying enzymes were purchased from New England Biolabs. Deoxyoligonucleotides (17-mers) complementary to both strands of carAB were synthesized with an ABI Model 380A DNA synthesizer.  $[\gamma^{-32}P]ATP$  (3000 Ci/mmol), deoxyadenosine 5'- $\alpha$ -[35S]thiotriphosphate (>1000 Ci/mmol), [U-14C]ADP (550 mCi/mmol), and adenosine 5'-O-(3-[35S]thiotriphosphate) (ATP $\gamma$ 35S)<sup>1</sup> (650 Ci/mmol) were obtained from Amersham. ATP, ATPγS, ADP, NADP, and NAD were purchased from Pharmacia. Diammonium carbamyl phosphate (Sigma) was converted to the potassium salt, and its purity (98.5%) was determined in the ornithine transcarbamylase reaction. Other reagents were of the highest quality available commercially.

Bacterial Strains. The genotypes and sources of the mutant and wild-type E. coli strains used in this study are presented in Table I. The mutant bacteriophage Plvir was used in all transductional crosses, as described by Miller (1972). The carB mutants L683 and L705 were obtained by transducing a wild-type strain of E. coli to tetracycline resistance with phage Pl<sub>vir</sub>(L895) that had been mutagenized with hydroxylamine according to the procedure of Hong and Ames (1971). TetR transductants were screened for arginine and uracil auxotrophs on minimal medium.

Western Analysis and Sucrose Gradient Centrifugation. Western analysis and sucrose gradient centrifugation of carbamyl phosphate synthetase in extracts of E. coli cells carrying mutations in carB were performed as previously described (Guillou et al., 1989).

Recombinant DNA Procedures. Standard protocols were used for isolation of chromosomal and plasmid DNA, digestion of DNA by restriction endonucleases, isolation of DNA fragments, ligation, and transformation of E. coli (Maniatis et al., 1982).

Cloning of Mutant carAB Genes. The carAB genes from mutants L683 (carB-43) and L705 (carB-117) were cloned by screening recombinant plasmid libraries prepared from chromosomal DNA of each strain. The chromosomal DNAs were digested with BgIII, and fragments ranging from 4.5 to 6.5 kb in size were isolated by preparative agarose gel electrophoresis and ligated to the BamHI site of pUC18. E. coli transformants were screened for the presence of the mutant carAB genes by colony hybridization. The two libraries yielded plasmids pLL43 and pLL117 each with the complete mutant carAB operon on a 5.6-kb Bg/II fragment. A similar method was used to clone a 5.6-kb Bg/II fragment containing the wild-type carAB operon from an isogenic strain.

DNA Sequencing. Plasmid DNA was isolated by the method of Birnboim and Doly (1979) and was further purified by cesium chloride equilibrium centrifugation. The nucleotide sequences of the mutant and wild-type carAB genes were obtained by the chain termination method (Sanger et al., 1987) using double-stranded plasmid DNA templates (Zhang et al., 1988) and a series of synthetic 17-mer oligodeoxynucleotides as primers.

Purification of Carbamyl Phosphate Synthetase. Wild-type carbamyl phosphate synthetase (specific activity, 3.6 µmol min<sup>-1</sup> mg<sup>-1</sup>, at 37 °C) was purified from E. coli L829 essentially as described by Rubino (1987). Mutant carbamyl phosphate synthetases were purified by the same procedure. except that the proteins were chromatographed on MonoQ columns (Pharmacia). The mutant and wild-type enzymes fractionated identically on sieving and ion-exchange columns. Enzyme purity was estimated to be about 98% by Coomassie staining of the proteins after SDS-polyacrylamide gel electrophoresis (Laemmli, 1970). Both wild-type and mutant enzyme preparations were determined to be free of contaminating ATPases, and of biotin-dependent carboxylases. The purified enzymes were desalted either by centrifugation through columns of Sephadex G-50 equilibrated with 0.1 M Tris-acetate (pH 7.6)/0.2 M sodium acetate/1 mM EDTA or by dialysis against 0.12 M potassium phosphate, pH 7.6, containing 1 mM EDTA.

Protein Determination. Protein concentration was determined from the absorbance at 280 nm, by using an extinction coefficient of 0.686/mg of protein (Anderson, 1977; Rubino et al., 1986).

Enzyme Assays. Carbamyl phosphate synthesis was determined by measuring the amount of citrulline formed in a coupled assay with ornithine transcarbamylase. Reaction mixtures contained, in a final volume of 0.10 mL, 0.05 M Hepes/NaOH, pH 7.6, 100 mM KCl, 5 mM ATP, 20 mM magnesium chloride, 20 or 200 mM NaHCO<sub>3</sub>, 20 mM glutamine, 5 mM ornithine, and 2 units of ornithine trans-

<sup>&</sup>lt;sup>1</sup> Abbreviations: ATPγS, adenosine 5'-O-(3-thiotriphosphate); Hepes, 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid; EDTA, ethylenediaminetetraacetic acid; CAD, multifunctional pyrimidine-specific enzyme comprising glutamine-dependent carbamyl phosphate synthetase, aspartate transcarbamylase, and dihydroorotase.

HCO<sub>3</sub><sup>-</sup>-dependent ATPase activity was determined by measuring [ $^{32}$ P]P<sub>i</sub> formed from [ $\gamma$ - $^{32}$ P]ATP. Reaction mixtures contained, in a final volume of 0.20 mL, 0.05 M Hepes/NaOH, pH 7.6, 100 mM KCl, 5 mM [ $\gamma$ - $^{32}$ P]ATP [(1-4) × 10<sup>6</sup> cpm/ $\mu$ mol], 20 mM magnesium chloride, and 20 or 200 mM NaHCO<sub>3</sub>. The reaction was started by addition of 5–20  $\mu$ g of desalted wild-type or mutant enzyme. After incubation for 10 min at 37 °C, the reaction was stopped by adding 1.0 mL of 0.1 N HCl containing acid-washed charcoal (50 mg/mL). After brief centrifugation, the radioactivity of an aliquot (0.70 mL) of the supernatant layer was determined by scintillation counting. Because [ $\gamma$ - $^{32}$ P]ATP is hydrolyzed nonenzymatically during the incubation, the experimental values were corrected for phosphate released in duplicate reactions carried out without added enzyme.

ATP synthesis from carbamyl phosphate and ADP was measured at 25 °C by following the increase in absorbance at 340 nm of NADPH generated in a coupled assay with hexokinase and glucose-6-P dehydrogenase. Each cuvette contained, in a total volume of 0.5 mL, 50 mM Hepes/NaOH, pH 7.0, 100 mM KCl, 5 mM ADP, 20 mM magnesium chloride, 10 mM dipotassium carbamyl phosphate (pH 6.5), 50 mM glucose, 1 mM NADP, 20 units of hexokinase, 2 units of glucose-6-phosphate dehydrogenase, and 6 mM  $P_i$  (added with the enzyme). The reaction was initiated by addition of 20  $\mu$ L of dialyzed wild-type (20  $\mu$ g) or mutant (100  $\mu$ g) enzyme.

Glutaminase activity was assayed in reaction mixtures containing 50 mM potassium phosphate, pH 7.6, and 20 mM glutamine, in a final volume of 0.20 mL. The reaction was started by addition of 25  $\mu$ g of wild-type or mutant protein. A control without enzyme was included for each assay. After incubation for 60 min at 37 °C, the reaction was terminated by addition of 20  $\mu$ L of 1 N HCl; after standing at 0 °C for 5 min, the solution was neutralized by addition of 25  $\mu$ L of 1 M Tris. The amount of glutamate formed was measured from the fluorescence of NADH after reaction with glutamate dehydrogenase and NAD (Brent & Bergmeyer, 1974).

Kinetic Analysis. The steady-state kinetics of the HCO<sub>3</sub>-dependent ATPase reaction were measured at 25 °C by coupling of the reaction to pyruvate kinase and lactate dehydrogenase and monitoring the disappearance of NADH at 340 nm. Each cuvette contained, in a final volume of 0.5 mL, 50 mM Hepes/NaOH, pH 7.6, 100 mM KCl, 20 mM magnesium chloride, 0.2 mM NADH, 1 mM phosphoenolpyruvate, 20 µg each of pyruvate kinase and lactate dehydrogenase, and varying concentrations [in the range (0.5- $5)K_{\rm m}$ ] of HCO<sub>3</sub><sup>-</sup> at different fixed concentrations of MgATP. When the concentration of NaHCO<sub>2</sub> in the assay mixtures exceeded 20 mM, the pH of the buffers was adjusted so that the final pH of the reaction mixtures was pH 7.5. In the case of the wild-type enzyme where the  $K_{\rm m}$  for HCO<sub>3</sub><sup>-</sup> is low, the actual concentration of HCO<sub>3</sub> in the assay was taken as the sum of the concentrations of the HCO<sub>3</sub> added and the endogenous  $HCO_3^-$  ( $\leq 0.4$  mM) present in the assay components. When ADP was added as a product inhibitor, the rate of ATP hydrolysis was determined at 25 °C by measuring the production of [<sup>32</sup>P]P<sub>i</sub> from [<sup>32</sup>P]ATP in duplicate assays with and without enzyme as described in the preceding section. Inhibition of ATP hydrolysis by ADP was determined with varying ATP and inhibitor concentrations at a fixed concentration (10 mM) of bicarbonate. Under these conditions, the kinetic data yielded a competitive pattern of inhibition, consistent with ADP being the last product released.

All of the kinetic data were analyzed according to the procedures described by Cleland (1979); constants and the standard error of their estimation were obtained by least-squares fit of initial rates to the appropriate equation, by using programs obtained from Dr. Ronald E. Viola (University of Akron, Akron, OH).

Binding Experiments. Binding of MgADP and MgATP<sub>\gamma</sub>S by wild-type and mutant carbamyl phosphate synthetases was determined at 4 °C by equilibrium dialysis, using dialysis cells (100-µL volume) and the methods described by Anderson (1977). In measurements of MgADP binding, the protein compartment contained either wild-type (42.2 µM) or mutant (58  $\mu$ M) carbamyl phosphate synthetase in a solution of 0.05 M Hepes/NaOH, 100 mM KCl, 10 mM potassium phosphate, 25 mM MgCl<sub>2</sub>, and 0.2 mM EDTA, pH 7.4. The ligand compartment contained [14C]ADP (~25000 cpm) and varying concentrations (20-1200  $\mu$ M) of ADP in the same buffer. In the MgATP<sub>\gamma</sub>S-binding experiments, the protein compartment contained wild-type (78.7  $\mu$ M) or Lys841 (75.9  $\mu$ M) mutant enzyme in 0.05 M Hepes/NaOH, 100 mM KCl, 10 mM potassium phosphate, 50 mM MgCl<sub>2</sub>, and 0.2 mM EDTA (pH 7.5). The ligand compartment contained ATP $\gamma^{35}$ S ( $\sim$ 40 000 cpm) and varying concentrations (0.05-2 mM) of ATP $\gamma$ S in the same buffer. The samples were dialyzed for 21 h at 4 °C. After equilibration, the radioactivity in the protein and ligand compartments was determined, and the amount of free and enzyme-bound ligand was calculated. All (>98%) of the radioactivity added initially was recovered at the end of the experiment. In control experiments, it was determined that ATP was not hydrolyzed to any significant extent during the course of the experiments. The binding constants n (number of sites) and K' (dissociation constant) were obtained after linear least-squares fitting of the binding data to the Scatchard equation.

Binding of UMP and ornithine was also determined in two different experiments by equilibrium dialysis. The wild-type or mutant enzyme (30 µM) in 50 mM Hepes/NaOH, 100 mM KCl, 10 mM potassium phosphate, and 0.2 mM EDTA, at pH 7.4, was dialyzed at 23 °C for 8 h against the same buffer containing [ $^{14}$ C]UMP (30  $\mu$ M) or [ $^{14}$ C]ornithine (100  $\mu$ M). The wild-type and Lys841 enzymes bound, respectively, 0.33 and 0.31 mol of [14C]UMP/mol of monomer. Both enzymes bound the same amount of [14C]ornithine, 0.18 ± 0.02 mol/mol of protein. When MgADP (10 mM) was included in the mixtures, binding increased to  $0.34 \pm 0.04$ mol/mol of enzyme monomer. From these measurements at a single concentration of the allosteric ligand, the dissociation constants of UMP and of ornithine were calculated to be 22 and 154  $\mu$ M, respectively. These values are comparable with the values of the dissociation constants previously reported by Anderson (1977), suggesting that the lysine substitution does not impair binding of these allosteric ligands.

#### RESULTS

Phenotype of carB Mutants. To identify mutations in catalytically important domains of the synthetase subunit, E. coli mutants auxotrophic for arginine and uracil were screened

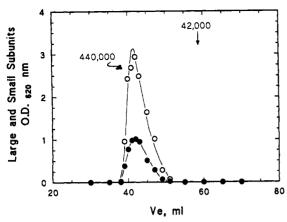


FIGURE 1: Elution profile of the  $\alpha$  and  $\beta$  subunits of Lys841 carbamyl phosphate synthetase upon chromatography on Superose 12. The purified mutant enzyme (0.94 mg) in 0.5 mL of 50 mM Hepes/ NaOH, pH 7.5, containing 0.1 M KCl, 10 mM potassium phosphate, and 0.5 mM EDTA was applied to a column (1.6  $\times$  50 cm) of Superose 12 equilibrated with the same buffer. The column was developed at 4 °C at a flow rate of 1 mL/min, and 1-mL fractions were collected. Samples (15  $\mu$ L) of each fraction were dissociated with SDS, and the subunits were separated in 9% polyacrylamide gels and stained with Coomassie Blue. The optical densities at 620 nm of the stained bands corresponding to the (•) glutaminase and (O) synthetase subunits were quantitated by densitometry and are expressed in arbitrary units. The protein concentration of the peak fraction was 0.103 mg/mL. The molecular weight of the carbamyl phosphate synthetase complex indicated in the figure was estimated from its elution volume  $(V_e)$  relative to standards of known molecular weight. The arrow indicates the elution volume corresponding to the molecular weight of the glutaminase subunit.

genetically and biochemically. Mutants with alleles mapping to the carB gene were analyzed as follows. Total cell extracts were first tested immunologically by the Western technique to exclude mutants with truncations or deletions in the synthetase subunit. Strains displaying normal concentrations of full-length synthetase and glutaminase subunits were further ascertained to have an  $\alpha\beta$  carbamyl phosphate synthetase complex by sucrose gradient centrifugation.

Two independent isolates, L683 and L705, had carbamyl phosphate synthetase with physical properties indistinguishable from those of the wild-type enzyme. Since both strains were determined to have the identical mutation, only the carB-117 mutation of L705 will be discussed. The results of analytical gel chromatography of the enzyme in L705, illustrated in Figure 1, indicate that both subunits of carbamyl phosphate synthetase coelute in a single peak. A similar elution pattern was observed with the wild-type carbamyl phosphate synthetase (data not shown). An average molecular weight of 440 000 is estimated for both wild-type and mutant enzymes. The salt and buffer conditions used for the gel filtration cause the enzyme to self-associate, and the estimated molecular weight is probably an average of an equilibrium mixture of monomer-dimer-tetramer. The two enzymes also have identical sedimentation properties in sucrose gradients. Both subunits cosediment as a single protein peak with an apparent molecular weight of about 160 000. The salt conditions in this analysis were chosen to minimize self-association of the enzyme and explain the lower molecular weight, which corresponds to the monomer. Neither method gave any indication of subunit dissociation nor was there any suggestion of a protein species with an elution volume  $(V_e)$  corresponding to the glutaminase subunit whose expected position in the elution profile is shown by the arrow in Figure 1. The carB-117 mutation, therefore, does not affect the ability of the synthetase and glutaminase subunits to form a stable complex.

Table II: Specific Activities of Wild-Type and Mutant Carbamyl Phosphate Synthetases in the Overall and Partial Reactions<sup>a</sup>

	sp act. (µmol	min-1 mg-1)
enzymatic act.	wild-type enzyme	Lys841 mutant enzyme
glutamine-dependent carbamyl phosphate synthesis	2.60	0
NH <sub>3</sub> -dependent carbamyl phosphate synthesis	1.81	0
glutaminase	0.025	0.020
HCO <sub>3</sub> -dependent ATPase		
20 mM HCO <sub>3</sub>	0.11	0.25
200 mM HCO <sub>3</sub> -	0.12	0.80
ATP synthesis from ADP and carbamyl phosphate	0.62	0.01

<sup>&</sup>lt;sup>a</sup> The assay conditions used to determine each of the enzyme activities are described under Materials and Methods.

The stability and normal physical behavior of the mutant enzyme coupled with its failure to catalyze any detectable synthesis of carbamyl phosphate with either glutamine or NH<sub>3</sub> as nitrogen donor are consistent with a mutation in a catalytic domain. In addition to the preliminary mapping data, several other lines of evidence also suggest a mutation in the synthetase component. L705 is unable to grow on minimal medium supplemented with 100 mM NH<sub>4</sub>Cl. This phenotype is most easily explained by a defect in the synthetase. E. coli mutants with a complete deletion of carA, for example, can synthesize sufficient carbamyl phosphate to support growth when supplied with high concentrations of ammonium ion in the medium. More significantly, transformation of the mutant with a specific region of carB yielded wild-type recombinants (see section below). Two other observations are worthy of note. The fact that the carB deletion strain C600 transformed with the mutant gene on high-copy plasmids fails to grow on minimal medium containing high NH<sub>4</sub><sup>+</sup> indicates a stringent mutation. Secondly, L705 has a reversion rate  $(1 \times 10^{-8})$  consistent with a single point mutation.

Sequence Analysis of carB-43 and carB-117 Mutations. Complementation of the mutant with a set of fragments of wild-type carAB on multicopy plasmids allowed the carB-43 and carB-117 mutations to be mapped to a 664 bp region between the SphI and SmaI sites of carB. The nucleotide sequences of this region of the mutant and the isogenic wildtype genes were obtained by the chain termination method (Sanger et al., 1987) with double-stranded DNA serving as templates for a set of oligodeoxynucleotide (17-mer) primers complementary to the sense strand of carB. The sequences indicate both mutants to have the identical  $G \rightarrow A$  transition at nucleotide 2521 of the carB reading frame. This single base pair substitution (GC for AT) changes the normal GAA codon of Glu841 to an AAA codon for lysine.

Properties of Mutant Carbamyl Phosphate Synthetase. The rates of overall carbamyl phosphate synthesis and of the partial reactions reflecting different steps of the mechanism are summarized in Table II for the wild-type and Lys841 enzymes. The mutant enzyme synthesizes carbamyl phosphate with a specific activity corresponding to less than 0.01% of wild type. Even though the glutaminase activity of the mutant enzyme appears to be normal, a more detailed kinetic analysis indicates this reaction to be affected by the Lys841 substitution as well (C. J. Lusty and M. Liao, unpublished results).

The enzymatic defects of the mutant enzyme were further characterized by assays of bicarbonate-dependent ATPase and transfer of the phosphoryl group from carbamyl phosphate to ADP, a reversal of the last step of the mechanism. The replacement of Glu841 with Lys has no apparent effect on the

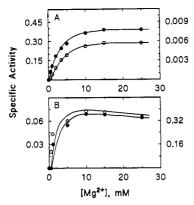


FIGURE 2: Dependence of ATP synthesis and HCO<sub>3</sub>-dependent ATPase on Mg<sup>2+</sup> concentration. In panel A, formation of ATP from carbamyl phosphate and ADP was measured at 25 °C, at pH 7.0, in reaction mixtures (0.5 mL) containing 50 mM Hepes/NaOH, pH 7.0, 50 mM KCl, 0.1 mM ADP, 30 mM potassium carbamyl phosphate, MgCl<sub>2</sub> as indicated, and either 15 µg of (O) wild-type or 112 μg of (•) Lys841 enzyme. In panel B, ATPase activity was assayed as ADP formed at 25 °C at pH 7.5 in a coupled assay with pyruvate kinase and lactate dehydrogenase. Each cuvette contained, in a final volume of 0.5 mL, 50 mM Hepes/NaOH, pH 6.8, 100 mM KCl, 100 mM NaHCO<sub>3</sub>, 0.1 mM ATP, MgCl<sub>2</sub> as indicated, and either 61.2  $\mu g$  of (O) wild-type or 11  $\mu g$  of ( $\bullet$ ) Lys841 enzyme. Left ordinate, specific activity of wild type; right ordinate, specific activity of Lys841 carbamyl phosphate synthetase.

Mg<sup>2+</sup> concentration dependence in either partial reaction. Both wild-type and mutant enzymes are saturated at the same concentration of the cation as shown by the normalized specific activities of the two enzymes in Figure 2. The HCO<sub>3</sub>-dependent ATPase activity is 2.5 times higher in the mutant than the wild-type carbamyl phosphate synthetase (Table II). A further 3-fold enhancement of this rate is seen in the mutant but not the wild-type complex at higher concentrations of HCO<sub>3</sub>. The Lys841 substitution has no effect on the pH dependence of the ATPase reaction which exhibits a broad optimum around pH 7.5.

In contrast to the nearly normal rate of HCO<sub>3</sub>-dependent ATPase in the mutant, synthesis of ATP from ADP and carbamyl phosphate occurs at only 1.6% of the wild-type rate. The differential loss of ATP synthesis but not ATPase activity suggests that the mutation affects predominantly the site involved in the phosphorylation of carbamate. The increased rate of ATP hydrolysis catalyzed by the Lys841 enzyme at the high concentrations of substrate, however, also indicates a second important effect of the mutation on the bicarbonate activation site in the amino-terminal half of the synthetase.

Binding Interactions of ADP and ATP $\gamma S$ . Binding of MgADP and of the analogue MgATP $\gamma$ S to carbamyl phosphate synthetase was determined from equilibrium dialysis experiments. The affinities of wild-type and Lys841 enzymes for each nucleotide were calculated from binding saturation curves obtained by measuring protein-bound ligand with different concentrations of ligand in the equilibration buffer. The apparent dissociation constants of the MgADP and MgATP<sub>\gammaS</sub> complexes derived from the titrations are summarized in Table III. The dissociation constants for both MgADP and MgATP $\gamma$ S are increased only by a factor of about 1.5-2. Scatchard plots were linear over the range of ligand concentrations used, suggesting that the dissociation constants for each nucleotide are similar at the two binding sites in the complex. The dissociation constant of ADP at the bicarbonate activation site was also determined kinetically as K, by assaying inhibition of the ATPase by ADP. The dissociation constants for ADP in the wild-type and mutant enzymes estimated by this means are almost identical to those

Table III: Binding of MgA	DP and MgA	$\Gamma P \gamma S^a$	_
binary complex	no. of sites, n <sub>app</sub>	dissociation constant, K <sub>i</sub> , 4 °C (mM)	$\Delta G_{\rm b}$ (kcal mol <sup>-1</sup> )
Glus 41 enzyme Ma ADP	19401	$0.11 \pm 0.01$	

biliary complex	Sites, napp	4 C (IIIVI)	moi )
Glu841 enzyme•MgADP	$1.8 \pm 0.1$	$0.11 \pm 0.01$	
Lys841 enzyme·MgADP	$1.9 \pm 0.1$	$0.23 \pm 0.02$	0.4
Glu841 enzyme-MgATP <sub>\gamma</sub> S	$1.5 \pm 0.1$	$0.42 \pm 0.06$	
Lys841 enzyme·MgATPγS	$1.5 \pm 0.1$	$0.64 \pm 0.10$	0.2

<sup>a</sup> The number of binding sites  $(n_{app})$  and the dissociation constants (Ki) at 4 °C were determined as described under Materials and Methods. Changes in the binding energies ( $\Delta G_b$ ) resulting from the amino acid replacement were calculated from the relationship  $\Delta G =$ RT ln  $[K_{i(mutant)}/K_{i(wild-type)}]$  (Fersht, 1985).

Table IV: Kinetic Parameters of Wild-Type and Lys841 Carbamyl Phosphate Synthetases in the ATP Synthesis Reaction<sup>a</sup>

kinetic parameter	wild-type enzyme	Lys841 enzyme
$k_{\rm cat}$ (s <sup>-1</sup> )	1.60	0.05
$K_{\rm m}$ for ADP ( $\mu$ M)	30	32
$k_{\rm cat}/K_{\rm m}$ for ADP (s <sup>-1</sup> M <sup>-1</sup> )	$5.3 \times 10^4$	1700
$K_{\rm m}$ for ADP (5 mM Orn) ( $\mu$ M)	6	25
$K_{\rm m}$ for carbamyl-P (mM)	0.90	10.8
$k_{\rm cat}/K_{\rm m}$ for carbamyl-P (s <sup>-1</sup> M <sup>-1</sup> )	1780	4.5

<sup>a</sup>Reaction conditions: 25 °C, pH 7.0, 50 mM Hepes/NaOH, 100 mM KCl, 15 mM free Mg<sup>2+</sup>, 30 mM potassium carbamyl phosphate (when MgADP was the varied substrate) or 5 mM MgADP (when carbamyl-P was the varied substrate), and 25  $\mu$ g of wild-type or 100  $\mu$ g of Lys841 carbamyl phosphate synthetase. Kinetic constants were obtained after fitting of the initial velocity data to the Michaelis-Menten

determined by equilibrium dialysis (cf. Table V below).

The Scatchard plot for the mutant complex extrapolated to about 1.5 mol of adenine nucleotide bound per mole of monomer. This value is the same as that seen in the wild-type complex. The difference in the binding constants corresponds to a decrease in the mutant enzyme of some 0.4 and 0.2 kcal mol<sup>-1</sup> in the binding energies of MgADP and MgATP $\gamma$ S, respectively. The results of these experiments argue against a direct involvement of Glu841 in nucleotide binding. The marginal effect of the mutation on the dissociation constants and the total amount of nucleotide bound suggests that replacement of Glu841 with lysine is unlikely to cause a serious perturbation in the conformation of the two adenine nucleotide binding domains. Measurements of ornithine and UMP binding also fail to reveal any major effects of Lys841 on the interaction of these allosteric effectors with the enzyme.

Kinetic Analysis of ATP Synthesis. As reported in Table II, a survey of the three easily measurable partial activities of carbamyl phosphate synthetase indicates a differential inactivation of the terminal ATP-dependent transphosphorylation reaction as a result of the Lys841 substitution. This reaction has been proposed to be catalyzed at a nucleotide-binding site located in the carboxyl half of the synthetase subunit (Post et al., 1990). To better understand the mechanistic basis for the lesion, a more detailed analysis of the catalytic properties of the mutant enzyme was carried out.

Because of its low ATP synthesis activity, the effect on the initial velocity of varying the concentration of either ADP or carbamyl phosphate at fixed but nonsaturating levels of the second substrate could not be determined accurately for the Lys841 enzyme. It was possible, however, to determine the steady-state constants  $k_{\text{cat}}$  (turnover number) and  $K_{\text{m}}$  by varying each substrate individually at saturating concentrations of the other. These values for the mutant and normal complexes are compared in Table IV. The turnover number is at least 30 times lower in the mutant enzyme.  $K_{\rm m}$  for ADP, on the other hand, is essentially unaffected. The 31-fold

reduction in the apparent second-order association constant  $k_{\rm cat}/K_{\rm m}$  for the binding of ADP is due almost entirely to the decrease in the rate constants for catalysis. These kinetic data indicate that the amino acid replacement does not strongly influence the binding interactions of ADP, a conclusion not inconsistent with the results of the equilibrium dialysis experiments which also exclude any compromising effect of the mutation on ADP binding.

A telling difference between the two enzymes with respect to the reverse reaction is the 12-fold higher  $K_{\rm m}$  for carbamyl phosphate in the mutant. Combined with the 30 times lower  $k_{cat}$ , it reduces the overall catalytic efficiency by about 400-fold. This dramatic decrease in  $k_{cat}/K_m$  explains, in part, the virtually complete inhibition of carbamyl phosphate synthesis. The 30 times lower turnover even in the presence of saturating carbamyl phosphate indicates that the principal effect of the Lys841 substitution is on the catalytic conversion of the enzyme-bound intermediates to carbamate and ATP in the backward reaction and inferentially to carbamyl phosphate and ADP in the forward synthetic direction. The higher  $K_{\rm m}$ and the decrease in  $k_{\rm cat}/K_{\rm m}$  also suggest a second effect of the amino acid replacement on binding of carbamyl phosphate. A consequence of a lower affinity for carbamyl phosphate might be a reduced affinity of the enzyme for carbamate as well, since the latter intermediate presumably binds to the same

Two other findings are of interest. Ornithine is a welldocumented allosteric activator of E. coli carbamylphosphate synthetase that acts by decreasing the  $K_m$  for ADP (Trotta et al., 1974a). The  $K_{\rm m}$  of the wild-type enzyme measured in the presence of ornithine is lowered 5-fold (Table IV). Although addition of ornithine causes a decrease in the  $K_m$  for ADP in the mutant, the change is small compared to the normal complex. Also significant is the absence of an effect on the reaction rate by the allosteric inhibitor UMP under conditions where the wild-type enzyme is inhibited by more than 95% (data not shown). These observations indicate that the changes in the active site induced by these allosteric molecules are no longer operative in the mutant enzyme.<sup>2</sup>

Coupling between HCO<sub>3</sub><sup>-</sup> Activation and Carbamate Phosphorylation. The results of the ATPase assays, while confirming the principal effect of the Glu841 → Lys841 substitution to be on phosphorylation of carbamate, also hinted at some perturbation of the catalytic site involved in carboxy phosphate synthesis previously mapped to the amino-terminal half of the synthetase subunit. This is evidenced in the faster rate of HCO<sub>3</sub>-dependent ATP hydrolysis, particularly at high concentrations of the anion (Table II).

This partial reaction was examined in more detail to elucidate how the mutation in the carboxyl domain affects the first step in carbamyl phosphate synthesis catalyzed in the amino half of the synthetase. The rates of ATP hydrolysis, a measure of carboxy phosphate synthesis, were determined as a function of ATP and HCO<sub>3</sub> concentrations in the assay. The initial velocities measured with the mutant complex in the presence of different concentrations of each substrate fit (with a value of  $\sigma$  equal to 0.016) the equation for the sequential mechanism:

$$v = VAB/K_AB + K_BA + AB + K_{i,A}K_B$$
 (5)

Table V: Kinetic Parameters in the HCO<sub>3</sub>-Dependent ATPase Reaction<sup>a</sup>

kinetic parameter	wild-type enzyme	Lys841 enzyme
$k_{\rm cat}$ (s <sup>-1</sup> )	$0.23 \pm 0.001$	$2.95 \pm 0.02$
$K_{\rm m}$ for ATP ( $\mu$ M)	$10 \pm 0.1$	$90 \pm 3$
$k_{\rm cat}/K_{\rm m}$ for ATP (s <sup>-1</sup> M <sup>-1</sup> )	$2.3 \times 10^4$	$3.3 \times 10^4$
$K_{i}$ for ATP ( $\mu$ M)	$97 \pm 1$	$158 \pm 6$
$K_{\rm m}$ for ${\rm HCO_3}^-$ (mM)	$0.64 \pm 0.01$	$59 \pm 1$
$k_{\rm cat}/K_{\rm m}$ for $HCO_3^-$ (s <sup>-1</sup> M <sup>-1</sup> )	360	50
$K_{iB}$ for $HCO_3^-$ (mM)	$5.9 \pm 0.1$	$103 \pm 6$
$K_i$ for ADP (mM <sup>-1</sup> )	$0.12 \pm 0.01$	$0.20 \pm 0.03$
$E_{\rm a}$ (kcal mol <sup>-1</sup> )	$17.3 \pm 0.1$	$14.2 \pm 0.2$

<sup>a</sup> Reaction conditions: 25 °C, pH 7.5, 0.05 M Hepes/NaOH, 100 mM KCl, 20 mM MgCl<sub>2</sub>, varying concentrations  $[(0.5-5)K_m]$  of NaHCO<sub>3</sub>, and different fixed concentrations  $[(0.5-4)K_m]$  of MgATP. The K<sub>i</sub> for ADP was determined as described under Materials and

suggesting that the Lys841 enzyme catalyzes HCO<sub>3</sub>-dependent ATP hydrolysis according to the same kinetic mechanism as wild type. The values of the kinetic constants obtained for the two enzymes are summarized in Table V. The turnover number for the ATPase reaction is 13 times greater in the mutant enzyme.  $K_{\rm m}$  for ATP is 10-fold higher. The ratio  $k_{\rm cat}/K_{\rm m}$ , equivalent to the apparent second-order rate constant  $(k_{1A})$  for addition of this substrate, is increased by 1.5-fold. A similar increase in the substrate inhibition constant  $K_{i,k}$  agrees with the binding data obtained with the ATP analogue ATP $\gamma$ S (Table III). With  $K_{i_A}$  and  $k_{1A}$ , we can calculate dissociation rates of 2.1 and 5.2 s<sup>-1</sup>. These values indicate that the mutation results in only a moderate increase in the rate constant for dissociation of ATP. In contrast,  $K_{\rm m}$ for HCO<sub>3</sub><sup>-</sup> is increased by 2 orders of magnitude. The resultant 7-fold lowering of  $k_{cat}/K_{mB}$  associated with the large increase in  $K_{i_B}$  indicates that the Lys841 substitution results in an enzyme that not only is less competent in binding carbamyl phosphate but also has an equally reduced affinity for  $HCO_3^-$ . The increased  $K_m$  for  $HCO_3^-$  implies that the lower binding is due to an increased dissociation of the enzymesubstrate complex E-ATP-HCO<sub>3</sub> as well as the enzyme-bound carboxy phosphate ADP intermediate.

A reasonable explanation for the observed increases in  $k_{cat}$ and  $K_{\rm m}$  for HCO<sub>3</sub> and, to a lesser extent, in  $K_{\rm m}$  for ATP and in K<sub>i</sub> for ADP is a destabilization of the Michaelis complexes E-ATP and E-ATP-HCO3 and of the ground-state complexes E-carboxy phosphate-ADP and E-ADP. Some degree of stabilization of the E-carboxy phosphate-ADP transition state may also contribute to the increase in  $k_{cat}$ . The changes in substrate binding energy due to the replacement of Glu841 can be estimated by comparing the relative changes in the dissociation constants  $K_{i_A}$  and  $K_{i_B}$  of the mutant and wild-type enzymes using the equation noted in Table III. The apparent decreases in the binding energies  $\Delta G_{\rm b}$  of ATP and of HCO<sub>3</sub> are calculated as 0.2 and 1.7 kcal mol<sup>-1</sup>, respectively. The lower binding energies of ATP and HCO<sub>3</sub> in the enzyme-substrate complex, by contributing a decrease of about 1.9 kcal mol-1 in the activation energy for  $k_{cat}$ , could account to a large extent for the higher reaction rate at saturating concentrations of substrate. A decrease in the activation energy for  $k_{cat}$  was substantiated experimentally when the mutant and wild-type enzymes were compared by measuring their Arrhenius energies of activation  $(E_a)$ . Over the temperature range 38-14 °C used in the experiments, Arrhenius plots of log V versus 1/K were linear with calculated slopes of  $-3798 \pm 30 \text{ K}^{-1}$  for wild-type enzyme and  $-3118 \pm 54 \text{ K}^{-1}$  for the mutant enzyme. From the derived values of  $E_a$  (Table V), we calculate the enthalpy of activation  $\Delta H^*(25\ ^{\circ}\text{C})$  for  $k_{\text{cat}}$  to be lower by about 3.1 kcal

<sup>&</sup>lt;sup>2</sup> Binding of either UMP or ornithine was measured at a single concentration of each effector by equilibrium dialysis. The calculated  $K_d$ values of 22 and 150  $\mu$ M, respectively, for UMP and ornithine are similar to those reported previously for wild-type enzyme. This would suggest that the lack of effect on the kinetics of the enzyme is not likely to be due to loss of binding of either UMP or ornithine.

Table VI: Effect of NH<sub>3</sub> on HCO<sub>3</sub>-Dependent P<sub>i</sub> Release from ATP<sup>a</sup>

	[ <sup>32</sup> P]P <sub>i</sub> rele min <sup>-1</sup>	
components added	wild-type enzyme	Lys841 enzyme
NaHCO <sub>3</sub> (20 mM)	0.10	0.20
NaHCO <sub>3</sub> (20 mM), NH <sub>4</sub> Cl (300 mM)	0.97	0.11
NaHCO <sub>3</sub> (200 mM)	0.12	0.72
NaHCO <sub>3</sub> (200 mM), NH <sub>4</sub> Cl (300 mM)	1.02	0.42

<sup>a</sup>Reaction conditions: 37 °C, pH 7.6, 50 mM Hepes/NaOH, 100 mM KCl, 4 mM [ $\gamma$ -<sup>32</sup>P]ATP (4 × 10<sup>6</sup> cpm/ $\mu$ mol), 20 mM MgCl<sub>2</sub>, and NaHCO<sub>3</sub> and NH<sub>4</sub>Cl as indicated.

mol<sup>-1</sup> as a result of the amino acid substitution.

Another novel property of the mutant carbamyl phosphate synthetase is the lack of stimulation of  $HCO_3$ -dependent ATPase by  $NH_4$ <sup>+</sup>. Normally, this activity is 10 times higher in the presence of ammonium ion (Table VI). In contrast, addition of  $NH_4$ <sup>+</sup> to the Lys841 enzyme reduces the ATPase activity. Since  $P_i$  release from ATP in the presence of ammonia is a measure of carboxy phosphate conversion to carbamate (eq 3), the failure of the cation to stimulate this partial reaction suggests that the Glu841  $\rightarrow$  Lys substitution has a further effect on formation of carbamate.

## DISCUSSION

The synthesis of carbamyl phosphate proceeds consecutively through two different enzyme-bound intermediates, carboxy phosphate and carbamate (Anderson & Meister, 1966). The mechanism by which the intermediates and carbamyl phosphate are formed has been outlined, but the nature of the catalytic site or sites responsible for catalyzing the individual steps is not well understood. Recently, site-directed mutagenesis has been successfully used to selectively inactivate two different domains in the synthetase subunit, one involved in activation of HCO<sub>3</sub><sup>-</sup> and the other in phosphorylation of carbamate to carbamyl phosphate (Post et al., 1990). While this functional dissection has helped to confirm earlier speculations about the locations of active-site domains in the amino and carboxyl halves of the synthetase component in the E. coli glutamine-dependent complex (Lusty et al., 1983), the topology of the two sites vis-à-vis each other is still unclear. Are the sites physically separated, necessitating a transfer of intermediates through a conduit similar to the indole channel

spanning the  $\alpha$  and  $\beta$  subunits of tryptophan synthase (Hyde et al., 1988), or are they part of a single complex structure?

Mutational analysis, particularly if the lesions in the enzyme are confined to specific steps of the overall reaction, not only can provide valid answers to questions of mechanism but also can clarify aspects of structure—function relationships. The  $E.\ coli$  mutant L705 reported in this paper is one of a large group of strains that were judged to have mutations in carB, based on their growth phenotypes and on genetic mapping data. L705(carB allele 2117) was selected for more detailed kinetic studies because it satisfied criteria generally recognized to indicate a mutation in an active site. Even though the mutation abolishes virtually all detectable carbamyl phosphate synthesis activity, it has no effect on the in vivo stability of the protein. Nor does it discernibly change the  $\alpha\beta$  structure and physical properties of the complex. Only 10 other mutants out of more than 250 tested exhibit these properties.

The carB-2117 allele of L705 was confirmed by sequencing to be in the synthetase component of the complex. The single base change in the sense strand from  $G \rightarrow A$  at nucleotide 2521 of the gene results in a substitution of a lysine for glutamic acid at residue 841 of the synthetase component. The amino acid change occurs within a highly conserved sequence context previously proposed to be an adenine-binding domain (Lusty et al., 1983) and more recently shown to function in phosphorylation of enzyme-bound carbamate (Post et al., 1990). These authors reported that substitution of Gly722 with isoleucine reduced phosphorylation of ADP by carbamyl phosphate but did not affect the ATPase activity of the enzyme. Glu841 located 120 amino acids away from Glv722 in the glycine-rich sequence of the adenine nucleotide binding fold is probably part of the same active-site domain. Glu841 is an invariant residue situated at the end of a cluster of hydrophobic amino acids (VYLIE) present in this domain in all carbamyl phosphate synthetases examined to date. The importance of the glutamic acid in catalysis is attested to by its presence in homologous domains of other enzymes such as pyruvate carboxylase, acetyl-CoA carboxylase, and carbamate kinase (Table VII). These enzymes can synthesize ATP from ADP and carbamyl phosphate (Attwood & Keech, 1984; Baur et al., 1989; Polakis et al., 1972) and in at least two cases (pyruvate and acetyl-CoA carboxylase) also catalyze an ATP-dependent activation of HCO<sub>3</sub><sup>-</sup> to form enzyme-bound carboxy phosphate (Attwood & Keech, 1984; Polakis et al., 1972). Each of these proteins has an active-site domain ho-

Table VII: Sequence Context of Glu841 in Carbamyl Phosphate Synthetases and Homologous Carboxylases<sup>a</sup>

enzyme/or	ganism	residues	amino acid sequence	ref
			G1u841	
CPS M CPS-P B CPS-A Yea CPS-P Yea CPS D CPSI Ra	ast melan.	827-860 818-851 846-880 1246-1279 1219-1252 1241-1274 1045-1078	N V O F A V K N N E - V Y L I E V N P R A A R T V P F V S K A T G V P N I O M A E K G G K - V Y V L E A N P R S S R T I V P F V S K A V G I P N I O F V L S O G E - V Y V L E V N P R S S R T I V P F L S K I T G I P N M O I I K D G E H T L K V I E C N I R A S R S F P F V S K V L G V N N I O F I A K D N E - I K V I E C N V R A S R S F P F I S K V V G V N N M O L I A K N N E - L K V I E C N V R A S R S F P F V S K T L D H D N V O F L V K G N D - V L V I E C N V R V S R S F P F V S K T L D H D N V O F L V K G N D - V L V I E C N V R V S R S F P F V S K T L D Y D N L O L I A K D D O - L K V I E C N V R V S R S F P F V S K T L G V D	b c d e f g h i
CK P.	aeruginosa	278-310	K D A V I G S L A D I V A I T E G K A G T R V S T R K A G I E Y R *	j
	ast icken	293-327 422-456	TAEFLVDNONRHYFIE INPRIQVEHTITEEITGID TVEYLYSODGSFYFLELNPRLOVEHPCTEMVADVN	k 1

<sup>&</sup>lt;sup>a</sup>The sequences are aligned to give maximal amino acid identities. Gaps represent postulated deletions/insertions. Invariant amino acid residues are boxed. Legend abbreviations: CPS, carbamyl phosphate synthetase; CPS-A, arginine-specific carbamyl phosphate synthetase; CPS-P, pyrimidine-specific carbamyl phosphate synthetase; CAD, carbamyl phosphate synthetase-aspartate transcarbamylase-dihydroorotase; PC, pyruvate carboxylase; ACAC, acetyl-CoA carboxylase; CK, carbamate kinase. <sup>b</sup>Nyunoya & Lusty (1983). <sup>c</sup>Morris & Reeve (1988). <sup>d</sup>C. Quinn and R. L. Switzer (personal communication). <sup>e</sup>Lusty et al. (1983). <sup>f</sup>Souciet et al. (1987). <sup>g</sup>Freund & Jarry (1987). <sup>h</sup>Nyunoya et al. (1985). <sup>f</sup>Simmer et al. (1990). <sup>f</sup>Baur et al. (1989). <sup>k</sup>Lim et al. (1988).

mologous to the carbamate phosphorylation domain of carbamyl phosphate synthetase with a conserved glutamic acid proximal to a stretch of hydrophobic amino acids. Also significant is that 5'-[p-(fluorosulfonyl)benzoyl]adenosine reacts with E. coli (Boettcher & Meister, 1980) and rat liver (Powers et al., 1983) carbamyl phosphate synthetases, and the carbamyl phosphate synthetase component of hamster CAD (Kim et al., 1991), causing complete loss of activity in each case. In studies of CAD (Kim et al., 1991), two of the tryptic peptides labeled by the ATP analogue originate from the carboxyl half of the carbamyl phosphate synthetase component. One of the peptides is an eight amino acid long fragment with the sequence VIENCRX. The glutamic acid residue in this peptide corresponds to Glu841 in the homologous E. coli carbamyl phosphate synthetase (cf. Table VII).

The aforementioned evidence indicates Glu841 to be an essential residue in a catalytic domain of the enzyme. This conclusion is also supported by the catalytic properties of the mutant enzyme. The replacement Glu841 → Lys confers a 10 000-fold lower rate of carbamyl phosphate synthesis. Since the assay is capable of detecting a single turnover of the enzyme, this corresponds to a virtually complete absence of activity. The dramatic loss of enzymatic competence in overall carbamyl phosphate synthesis can be rationalized in terms of distinct effects on both carbamate phosphorylation and HCO<sub>3</sub> activation. Qualitatively and quantitatively, the more damaging effect of the mutation is seen in the new kinetic parameters describing the terminal part of the catalytic cycle. Phosphorylation of ADP by carbamyl phosphate, a partial reaction reflecting reversal of the last step of the mechanism, is reduced by more than 2 orders of magnitude. The lowered  $k_{\rm cat}$  and higher  $K_{\rm m}$  for carbamyl phosphate in this reaction suggest less efficient conversion of the transition-state intermediate E-ADP-carbamyl phosphate to E-ATP-carbamate.

Among the possible functions of Glu841 are (1) binding of ATP through Mg<sup>2+</sup>, (2) formation of a phosphorylated enzyme intermediate, (3) extraction of a proton from carbamic acid in preparation for a nucleophilic attack on ATP, and (4) provision of an electronically favorable local environment for the removal of a highly negatively charged product. Of these, only phosphorylation of the glutamic acid residue can be discounted in view of the absence of any evidence for the participation of a phosphorylated intermediate in the catalytic mechanism (Meister, 1989). The involvement of glutamic acid in adenine nucleotide binding through Mg<sup>2+</sup>, either directly or through a water molecule, seems unlikely in view of the ability of the mutant enzyme to bind ADP with an affinity similar to wild type. At present, we favor the notion that Glu841 may participate in catalysis by acting as a protonwithdrawing (forward reaction) or -donating group (reverse reaction). The large decrease in the  $k_{cat}$  of the mutant enzyme for overall carbamyl phosphate synthesis is consistent with a role of Glu841 as a general base or acid. A similar role of aspartic acid residues has been implicated in phosphoryl group transfer catalyzed by hexokinase (Viola & Cleland, 1978) and phosphofructokinase (Hellinga & Evans, 1987; Shirakihara & Evans, 1988). The hydrolysis of GTP promoted by ras p21 also occurs by a general base type of mechanism dependent on a glutamine residue (Pai et al., 1990).

A second factor contributing to the absence of overall carbamyl phosphate synthesis is the change in the kinetic parameters of HCO<sub>3</sub><sup>-</sup> activation. At saturating concentrations of  $HCO_3^-$ , the  $K_m$  for ATP in this reaction is increased 10-fold; however, the magnitude of the increase in  $K_{\rm m}$  corresponds almost entirely to the increase in  $k_{cat}$ , indicating that the lysine

substitution has little effect on the rate of association of ATP, but increases one or more of the rate constants for ATP and bicarbonate conversion to ADP and carboxy phosphate. These results tend to argue against any important role of Glu841 in binding of ATP in the initial activation step or in any direct involvement of Glu841 in the formation of carboxy phosphate. Despite the 10 times higher HCO<sub>3</sub>-dependent ATPase activity assayed in the presence of saturating concentrations of both substrates, the overall catalytic efficiency of the bicarbonate activation site is 7-fold lower in the mutant enzyme due to the 100-fold increase in  $K_{\rm m}$  for bicarbonate and an increased dissociation of enzyme-bound carboxy phosphate.

The altered kinetic parameters of the ATPase reaction indicate that the mutation also affects this catalytic step. The reaction of HCO<sub>3</sub> with ATP has been demonstrated to be catalyzed by a domain located in the amino-terminal half of the synthetase (Post et al., 1990). The apparent functional coupling of two physically separated domains in the protein, one responsible for bicarbonate activation and the other for phosphorylation of carbamate, suggests that they may be part of a single structural unit with amino acid residues recruited from both domains. According to this interpretation, the Glu841 → Lys841 replacement could be visualized not only to lead to the loss of a critical catalytic residue but also to induce other more subtle distortions in a local configuration of residues necessary for stabilizing the enzyme-bound intermediates. The composition of such a complex site in fact might also include side chains originating in the glutaminase component. In this context, a Cys269 → Ser substitution in the glutaminase subunit that abolishes the hydrolysis but not the binding of glutamine also results in stimulation of the HCO<sub>3</sub>-dependent ATPase activity of the synthetase. This enhancement in ATPase has been shown to be a consequence of destabilization and increased dissociation of the carboxy phosphate intermediate (Mullins et al., 1991).

An important aspect of the mechanism and one thus far neglected in this discussion is the conversion of carboxy phosphate to carbamate. In the presence of NH<sub>3</sub>, carbamate and P<sub>i</sub> are formed in equimolar amounts (eq 3), and the HCO<sub>3</sub>-dependent P<sub>i</sub> release from ATP has been generally assumed to be a measure of carbamate formation. NH<sub>3</sub> normally causes the rate of P<sub>i</sub> formation to be stimulated by a factor of 10-15 (Table VI). In contrast to wild-type carbamyl phosphate synthetase, addition of NH<sub>3</sub> to the Lys841 enzyme fails to elicit any increase in Pi production. This observation suggests an impairment in the mutant of carbamate formation from carboxy phosphate and NH3 as well (eq 3). The apparent inability of the mutant carbamyl phosphate synthetase to convert carboxy phosphate to carbamate further points to a possible role of Glu841 in this reaction as well.

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# REFERENCES

Anderson, P. M. (1977) Biochemistry 16, 587-592.

Anderson, P. M., & Meister, A. (1965) Biochemistry 4, 2803-2808

Anderson, P. M., & Meister, A. (1966) Biochemistry 5, 3157-3163.

Anderson, P. M., & Carlson, J. D. (1975) Biochemistry 14, 3688-3694.

Attwood, P. V., & Keech, D. B. (1984) Curr. Top. Cell. Regul.

- Baur, H., Luethi, E., Stalon, V., Mercenier, A., & Haas, D. (1989) Eur. J. Biochem. 179, 53-60.
- Birnboim, H. C., & Doly, J. (1979) Nucleic Acids Res. 7, 1513-1523.
- Boettcher, B. R., & Meister, A. (1980) J. Biol. Chem. 255, 7129-7133.
- Bolivar, F., & Backman, K. (1979) Methods Enzymol. 68, 245-267.
- Brent, E., & Bergmeyer, H. U. (1974) in *Methods of Enzymatic Analysis* (Bergmeyer, H. U., Ed.) pp 1704-1708, Academic Press, New York.
- Cleland, W. W. (1979) Methods Enzymol. 63, 103-138.
- Csonka, L. N., & Clark, A. J. (1980) J. Bacteriol. 143, 529-530.
- Fersht, A. (1985) in *Enzyme Structure and Mechanism*, 2nd ed., pp 303, 350, W. H. Freeman, New York.
- Foley, R., Poon, J., & Anderson, P. M. (1971) *Biochemistry* 10, 4562-4569.
- Guillou, F., Rubino, S. D., Markovitz, R. S., Kinney, D. M., & Lusty, C. J. (1989) Proc. Natl. Acad. Sci. U.S.A. 86, 8304-8308.
- Hellinga, H. W., & Evans, P. R. (1987) Nature (London) 327, 437-439.
- Hong, J. S., & Ames, B. N. (1971) Proc. Natl. Acad. Sci. U.S.A. 68, 3158-3162.
- Hyde, C. C., Ahmed, S. A., Padlan, E. A., Miles, E. W., & Davies, D. R. (1988) J. Biol. Chem. 263, 17857-17871.
- Kim, H., Lee, L., Evans, D. R., & Caplow, M. (1968) J. Am. Chem. Soc. 90, 6795-6803.
- Laemmli, U. K. (1970) Nature (London) 227, 680-685.
- Lusty, C. J., Widgren, E. E., Broglie, K. E., & Nyunoya, H. (1983) J. Biol. Chem. 258, 14466-14477.
- Maniatis, T., Fritsch, E. F., & Sambrook, J. (1982) in *Molecular Cloning: A Laboratory Manual*, Cold Spring Harbor Laboratory, Cold Spring Harbor, New York.
- Matthews, S. L., & Anderson, P. M. (1972) *Biochemistry* 11, 1176–1183.
- Meister, A. (1989) Adv. Enzymol. Relat. Areas Mol. Biol. 62, 315-374.
- Mergeay, M., Gigot, D., Beckmann, J., Glansdorff, N., & Pierard, A. (1974) Mol. Gen. Genet. 133, 299-316.

- Miller, J. H. (1972) in Experiments in Molecular Genetics, Cold Spring Harbor Laboratory, Cold Spring Harbor, NY.
- Mullins, L. S., Lusty, C. J., & Raushel, F. M. (1991) J. Biol. Chem. 266, 8236-8240.
- Nyunoya, H., & Lusty, C. J. (1984) J. Biol. Chem. 259, 9790-9798.
- Pai, E. F., Krengel, U., Petsko, G. A., Goody, R. S., Kabsch, W., & Wittinghofer, A. (1990) EMBO J. 9, 2351-2359.
- Pierard, A., Glansdorff, N., Mergeay, M., & Wiame, J. M. (1965) J. Mol. Biol. 14, 23-36.
- Pinkus, L. M., & Meister, A. (1972) J. Biol. Chem. 247, 6119-6127.
- Polakis, S. E., Guchhait, R. B., & Lane, M. D. (1972) J. Biol. Chem. 247, 1335-1337.
- Post, L. E., Post, D. J., & Raushel, F. M. (1990) J. Biol. Chem. 265, 7742-7747.
- Powers, S. G., Muller, G. W., & Kafka, N. (1983) J. Biol. Chem. 258, 7545-7549.
- Raushel, F. M., & Villafranca, J. J. (1980) *Biochemistry* 19, 3170-3174.
- Rubino, S. D., Nyunoya, H., & Lusty, C. J. (1986) J. Biol. Chem. 261, 11320-11327.
- Rubino, S. D., Nyunoya, H., & Lusty, C. J. (1987) J. Biol. Chem. 262, 4382-4386.
- Sanger, F., Miklen, S., & Coulson, A. R. (1987) Proc. Natl. Acad. Sci. U.S.A. 84, 4767-4771.
- Shirakihara, Y., & Evans, P. R. (1988) J. Mol. Biol. 204, 973-994.
- Trotta, P. P., Burt, M. E., Haschemeyer, R. H., & Meister, A. (1971) *Proc. Natl. Acad. Sci. U.S.A.* 68, 2599-2603.
- Trotta, P. P., Estis, L. F., Meister, A., & Haschemeyer, R. H. (1974a) J. Biol. Chem. 249, 482-491.
- Trotta, P. P., Pinkus, L. M., Haschemeyer, R. H., & Meister, A. (1974b) J. Biol. Chem. 249, 492-499.
- Viola, R. E., & Cleland, W. W. (1978) Biochemistry 17, 4111-4117.
- Wimmer, M. J., Rose, I. A., Powers, S. G., & Meister, A. (1979) J. Biol. Chem. 254, 1854-1859.
- Zhang, H., Scholl, R., Browse, J., & Somerville, C. (1988) Nucleic Acids Res. 16, 1220.